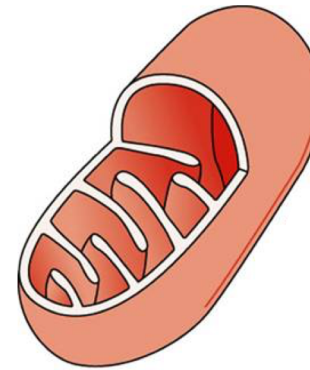


We will discuss:

1. Cells and Organs
2. Chemical Components of Cells
3. Proteins, Enzymes
- 4. Energy, Metabolism**
5. DNA, Chromosomes, Replication
6. Gene expression
7. Recombinant Techniques
8. Membrane and Transport
9. Intracellular Trafficking
10. Cytoskeleton
11. Cell division, Mitosis
12. Genetics, Meiosis
13. Cell communication, Signaling
14. Tissue, Tissue regeneration

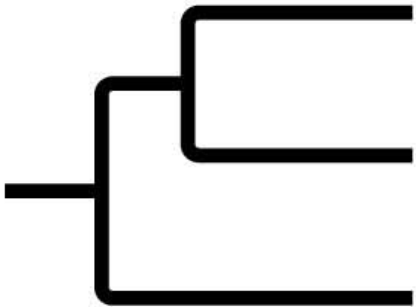
Key organelle : mitochondrion



Mitochondrion
(ATP synthesis)

MAJOR THEMES IN BIOLOGY

Evolution



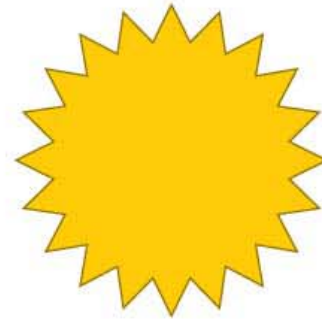
**Structure/
Function**



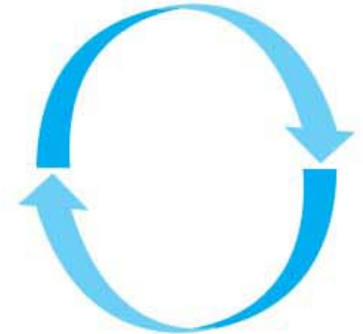
**Information
Flow**



**Energy
Transformations**



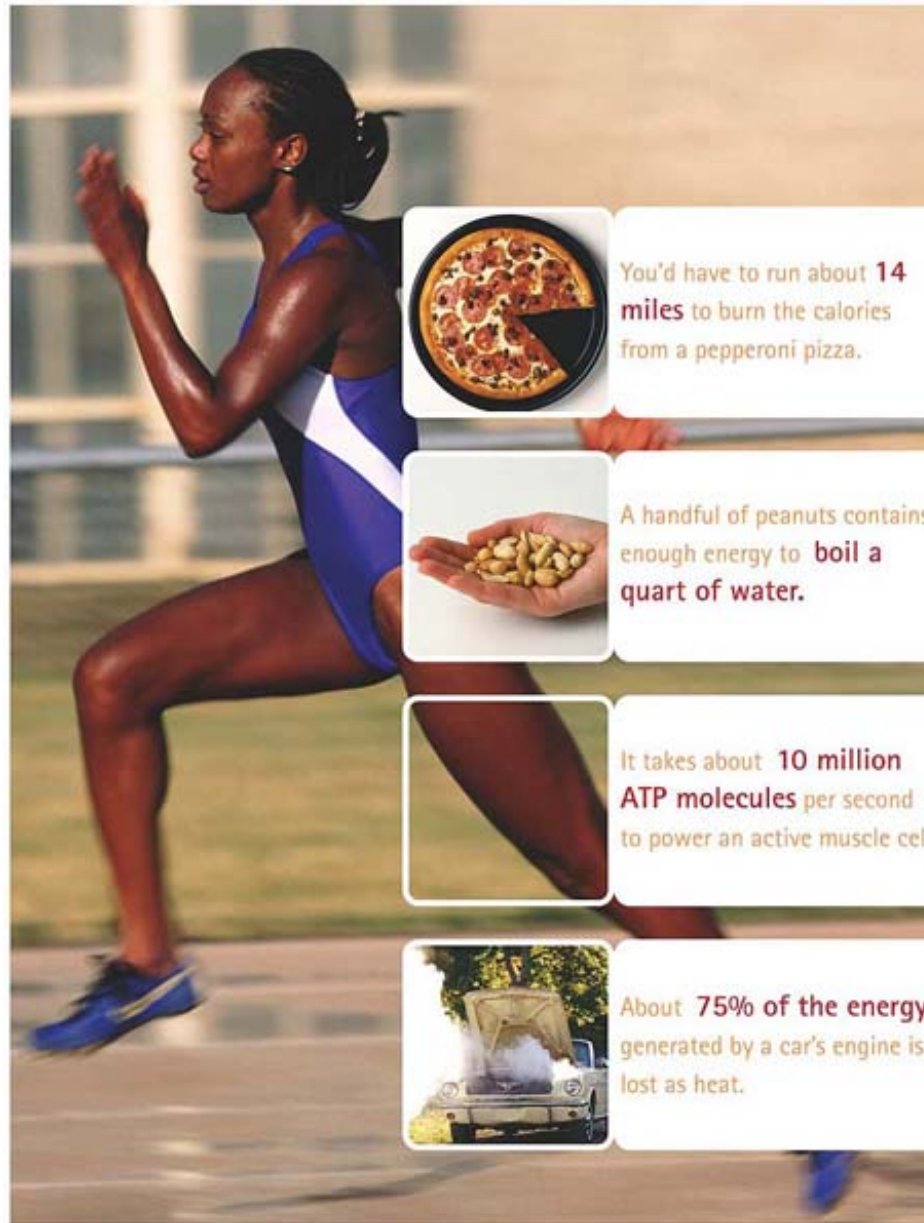
**Interconnections
within Systems**



Today

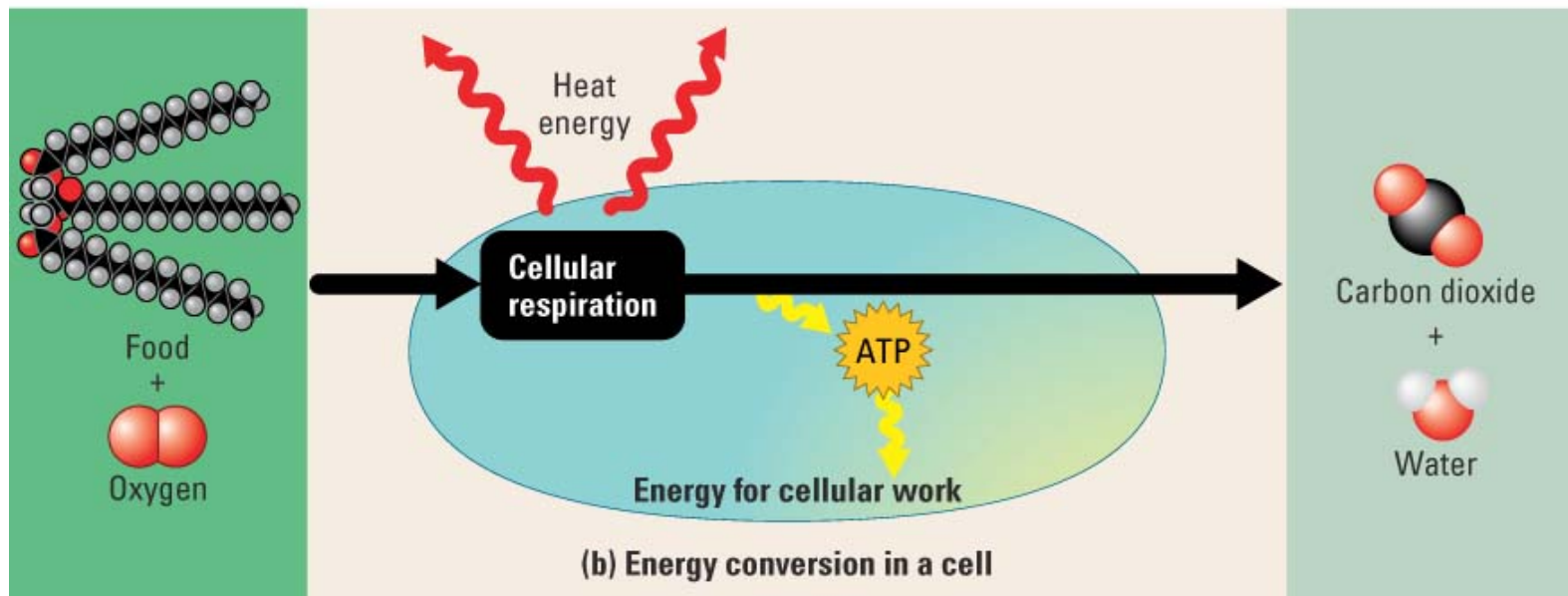
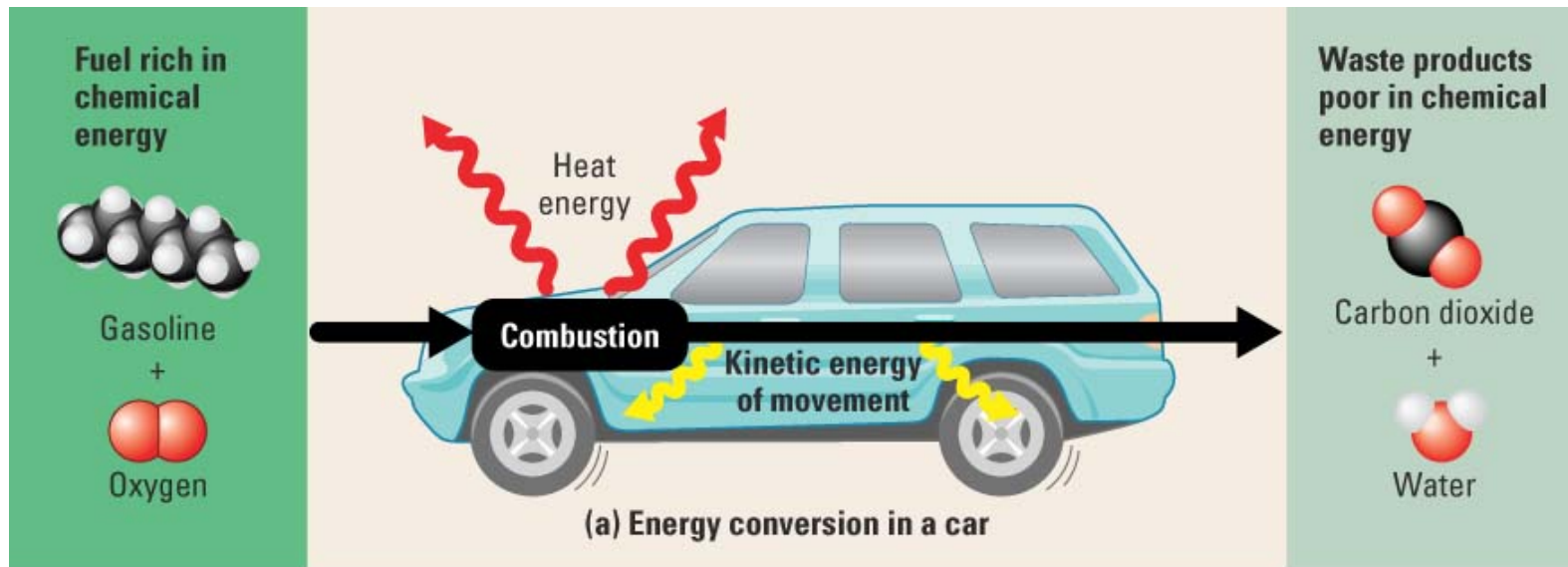
Essential Biology :

Chapter 5



Per day, it takes as much ATP as your own body weight to power your life.

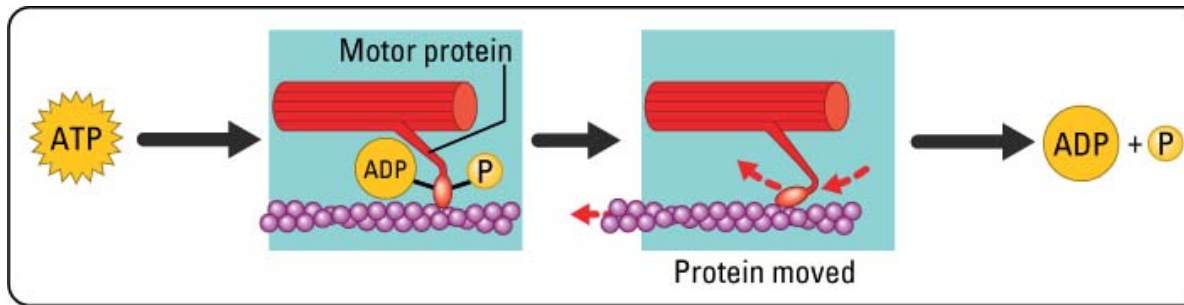
Efficiency



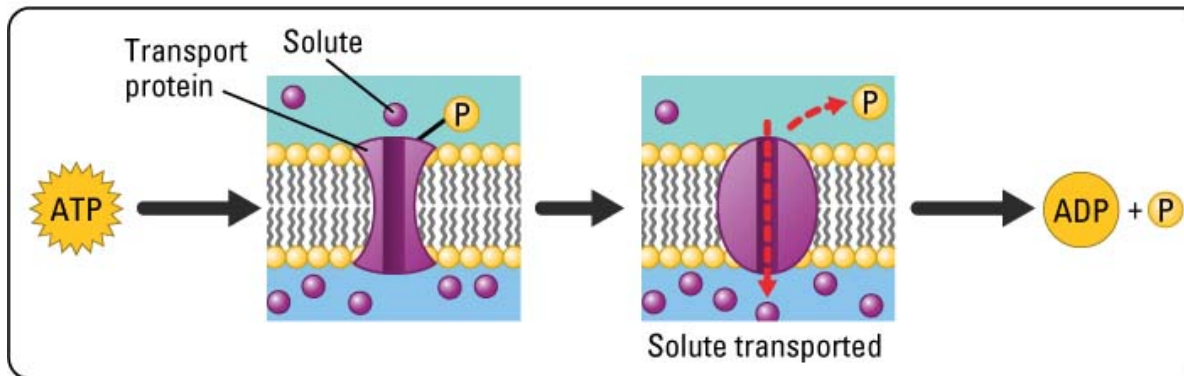
Energy for mechanical work.



Obvious !!

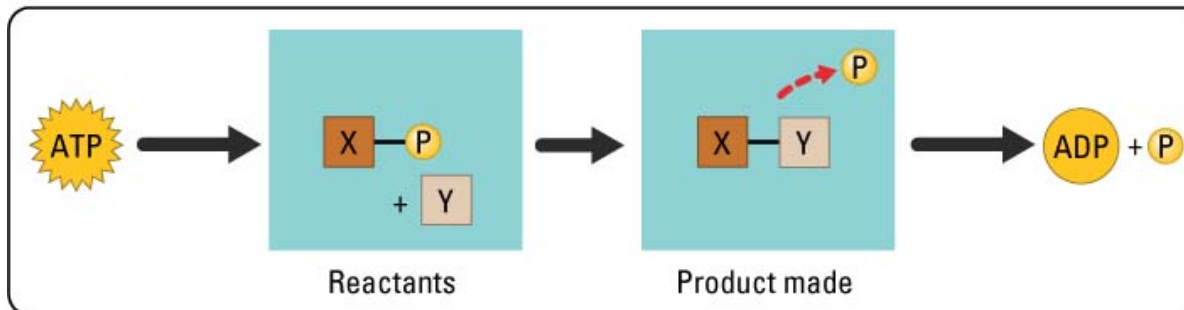


(a) Mechanical work



(b) Transport work

Think about
resting membrane potential



(c) Chemical work

Think about
assembling monomers
into polymers :
amino acids → proteins
nucleotides → nucleic acids

Chapter 6

Cellular Respiration: Obtaining Energy from Food



Two ways to synthesize ATP

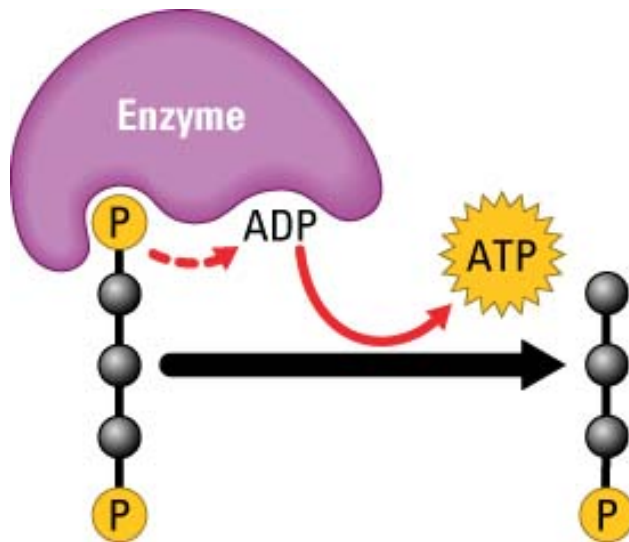
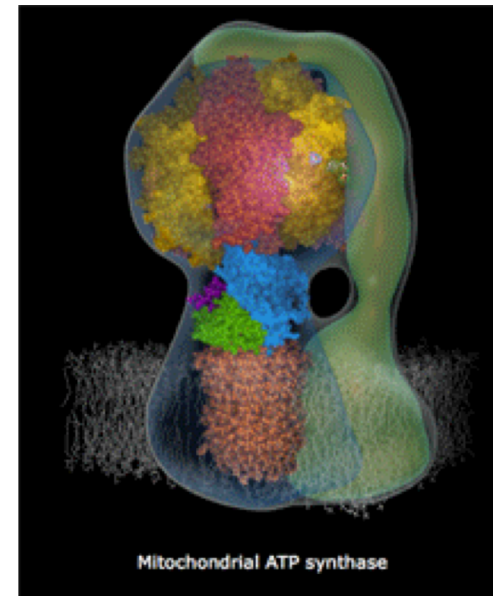


Figure 6.9 ATP synthesis by direct phosphate transfer. Glycolysis generates ATP when enzymes transfer phosphate groups directly from fuel molecules to ADP.



Chapter 7

Photosynthesis: Using Light to Make Food

Energy considered at
the biosphere organization level



The Alimentary Canal

The alimentary canal (or gut): food passes through and is processed in various ways

1. Mechanical digestion

→ teeth

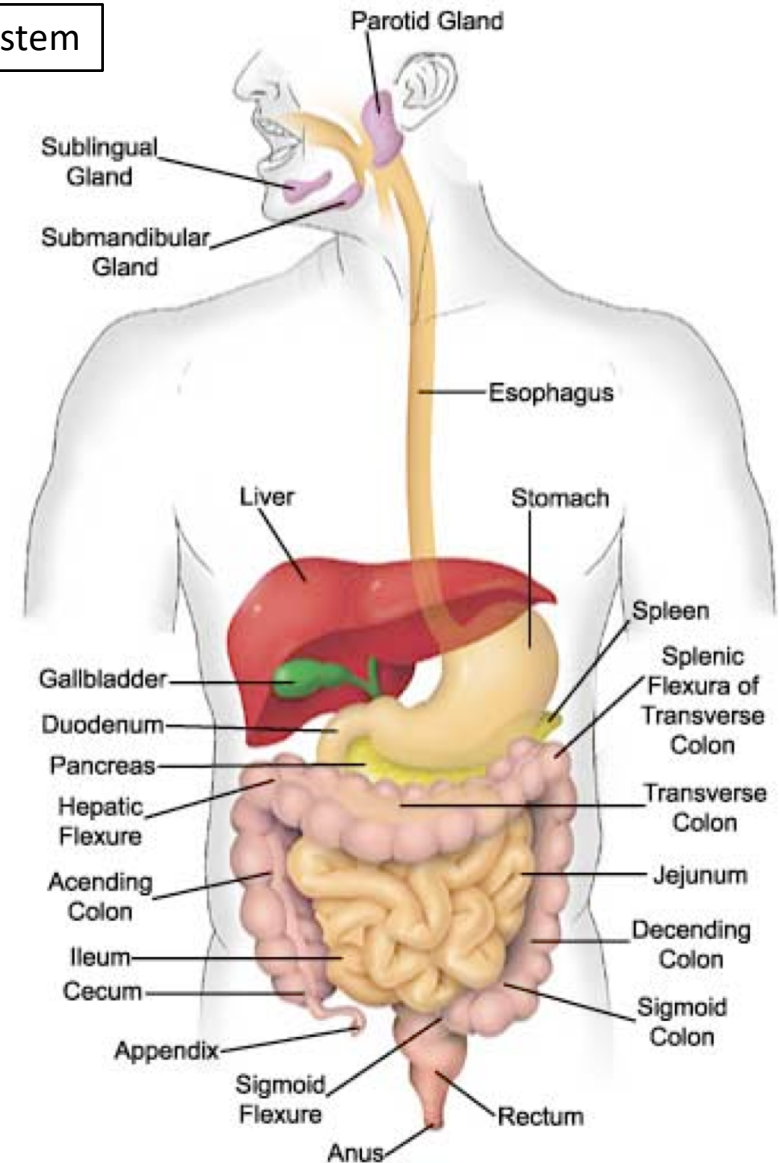
2. Chemical digestion

→ Enzymes: breakdown of macromolecules (stomach)

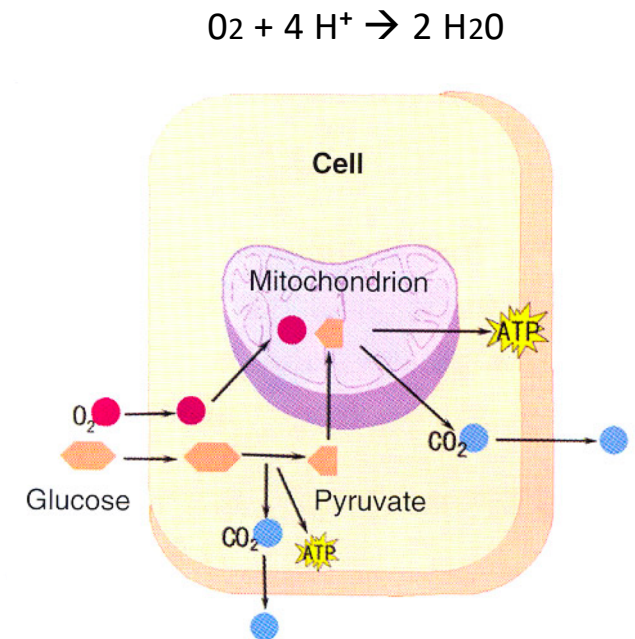
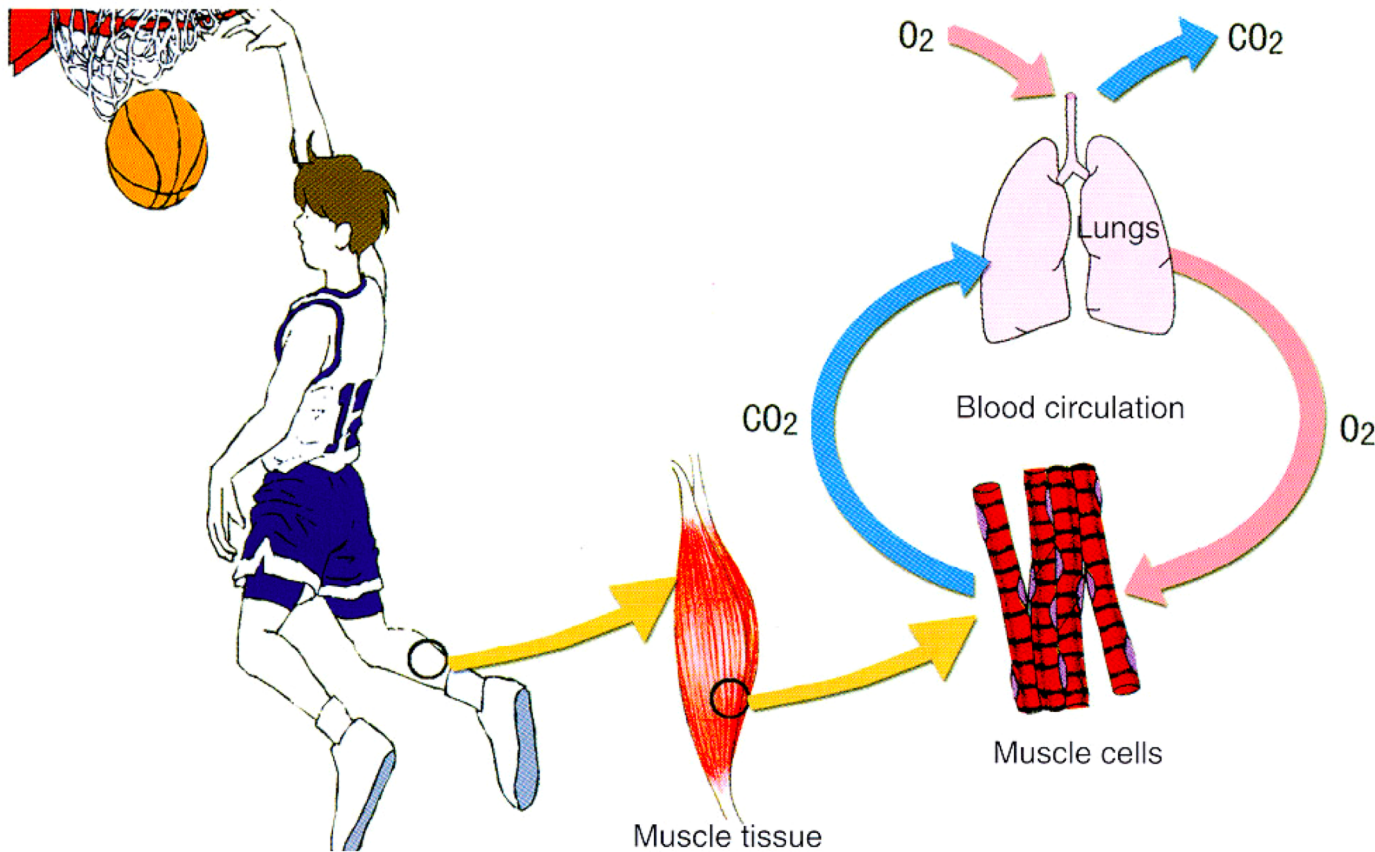
→ absorption of small unitary molecules (small intestine)

→ delivery to the blood stream

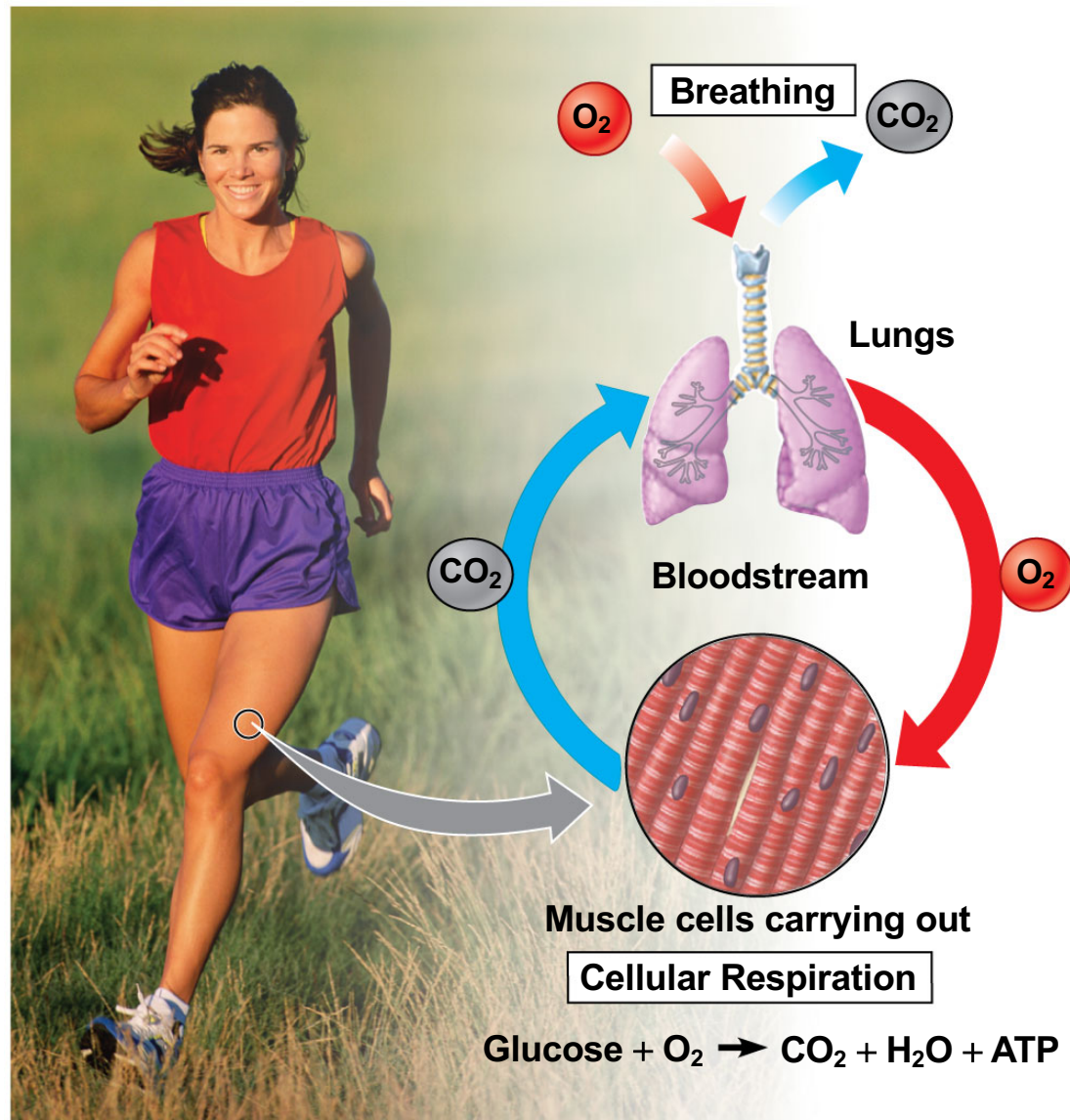
Organization level : system



Aerobic metabolism consumes O_2 and produces CO_2

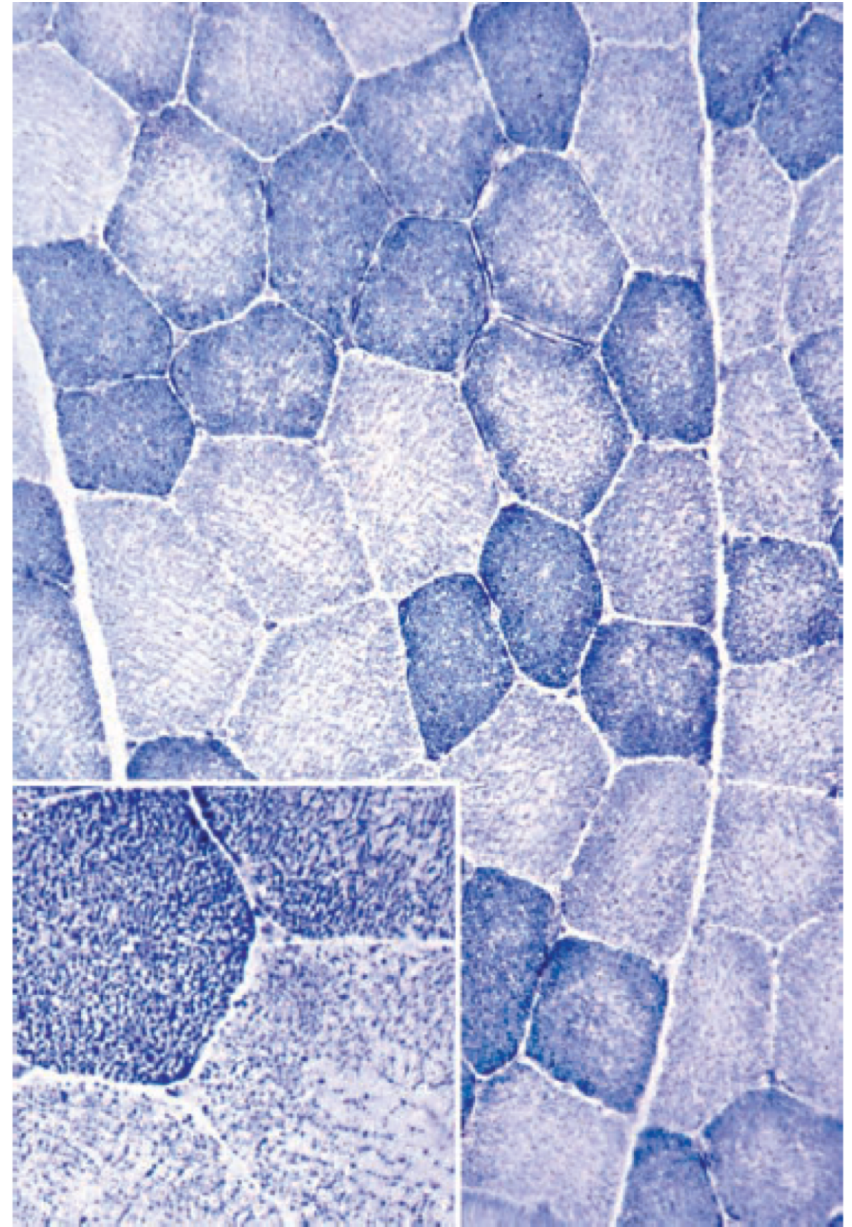
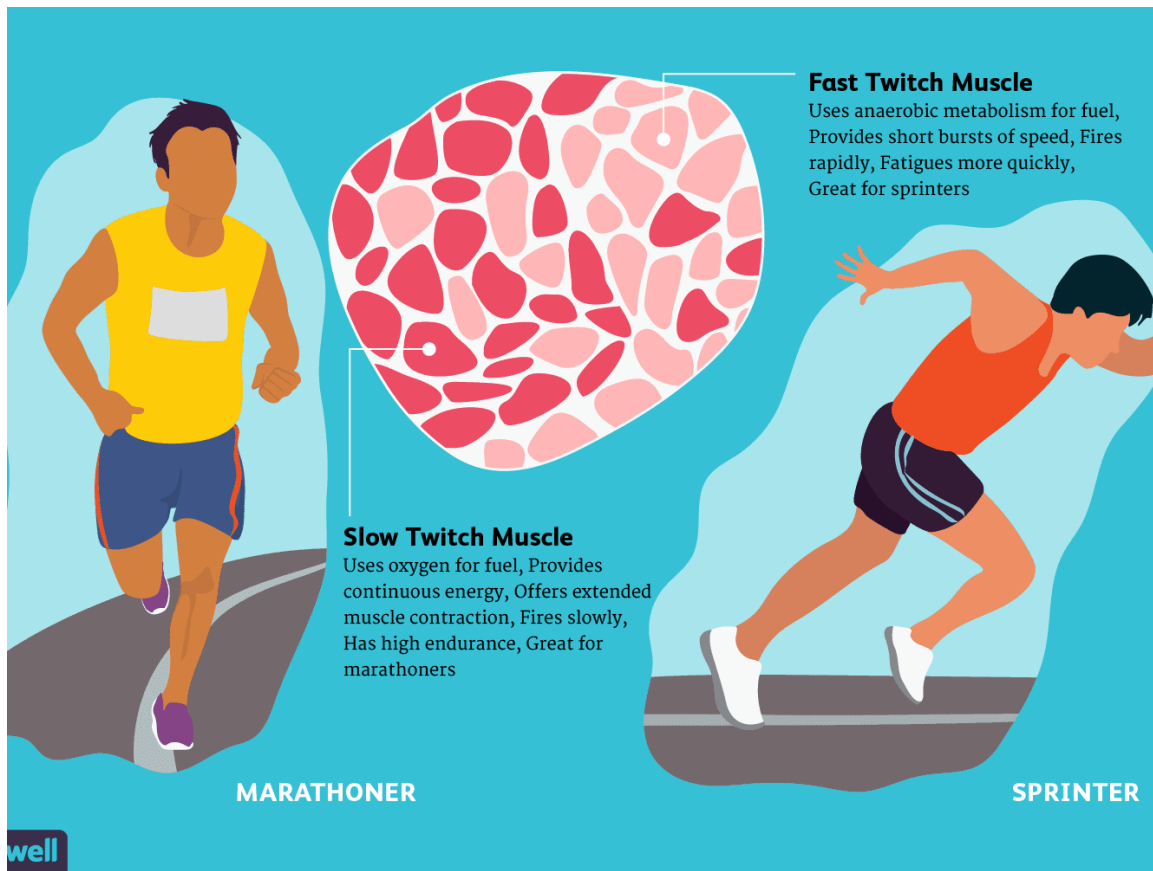


Energy



Two types of muscle fibers

Some muscle fibers have a lot of mitochondria.



Link between photosynthesis and respiration

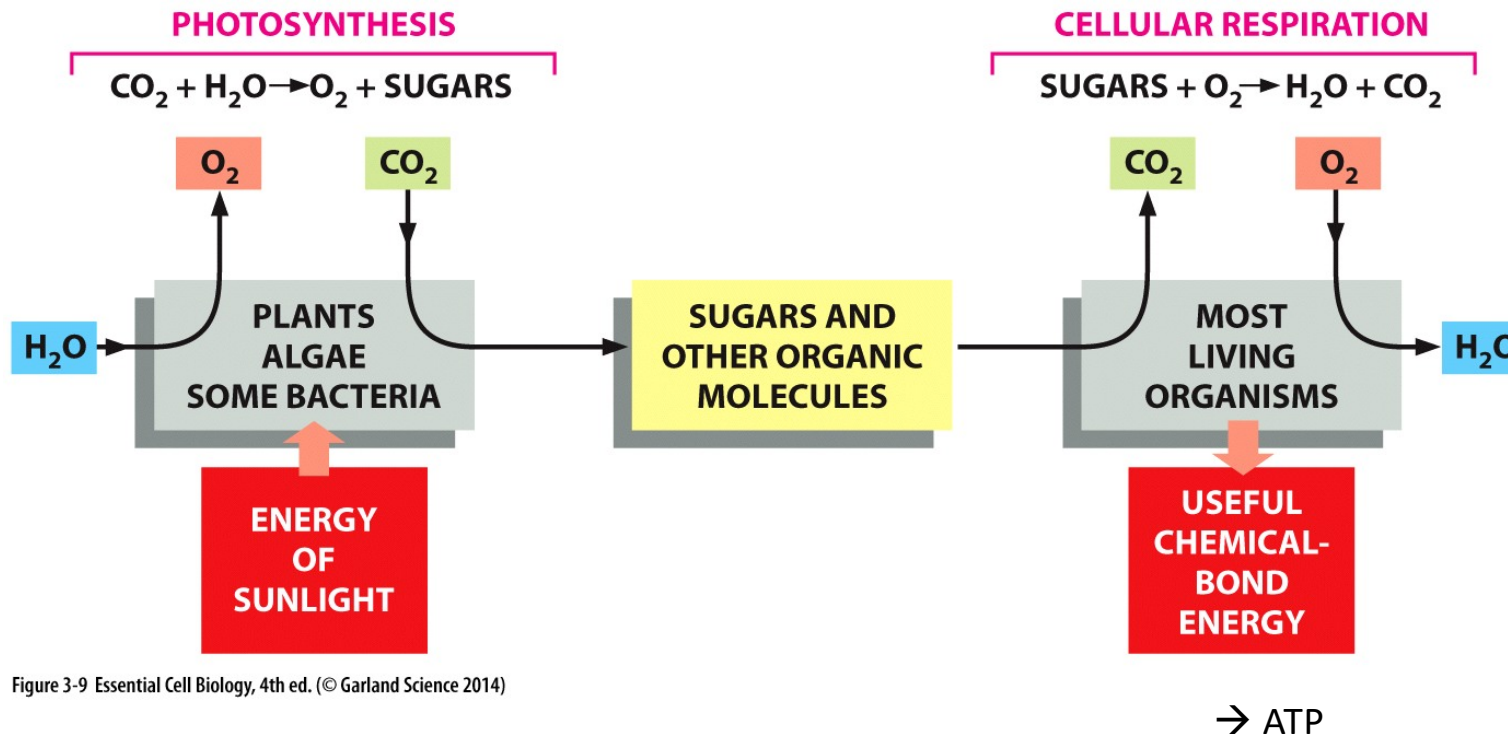
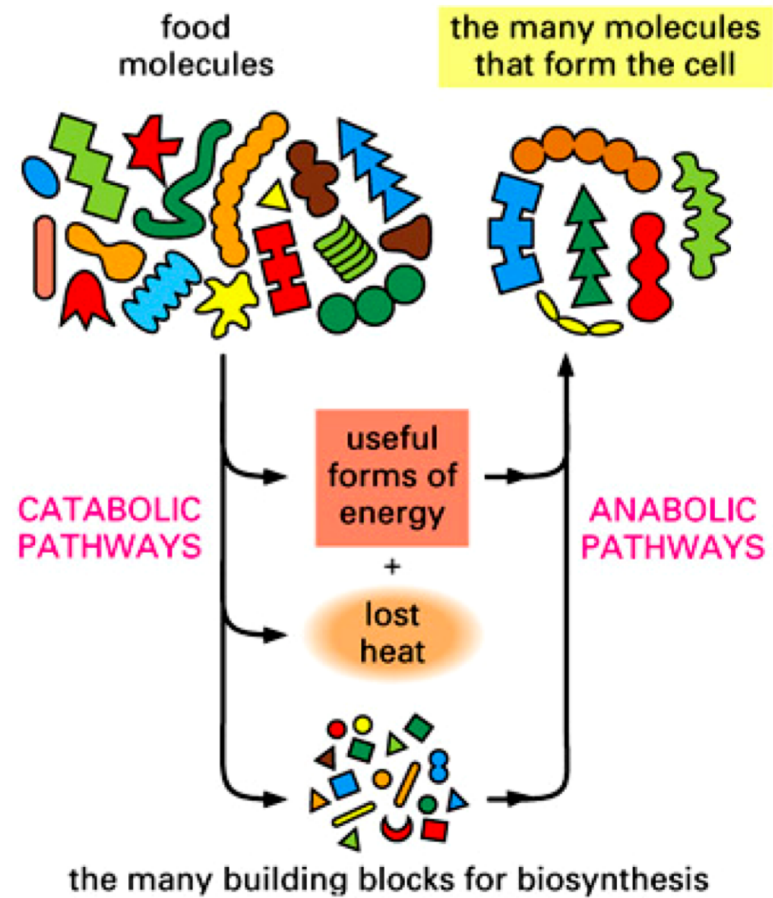
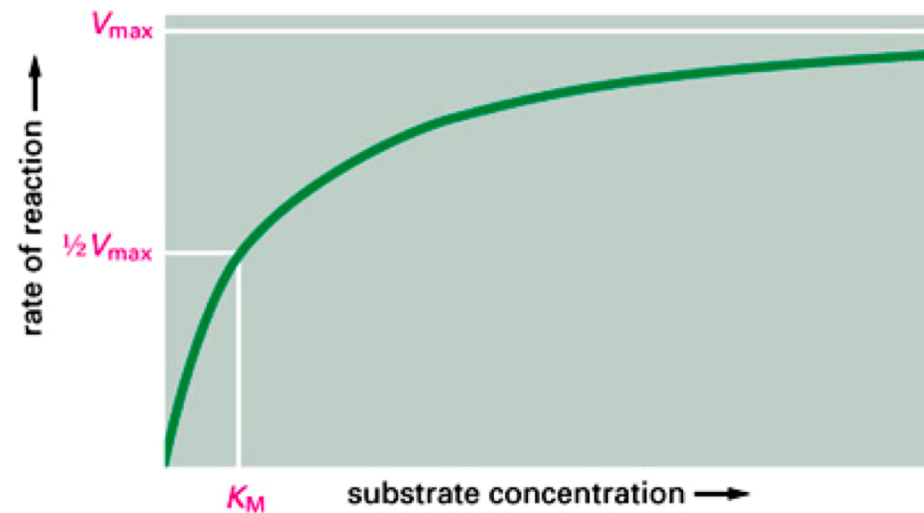
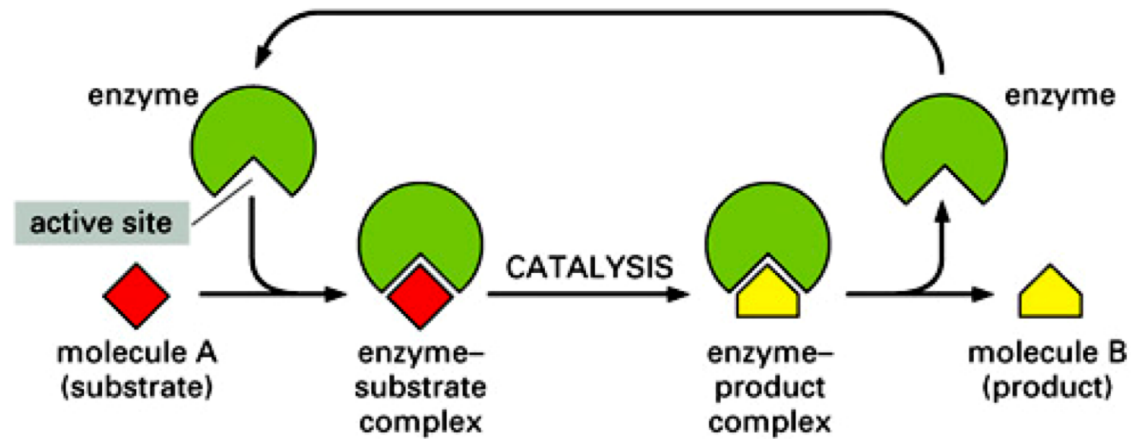


Figure 3-9 Essential Cell Biology, 4th ed. (© Garland Science 2014)

Catabolic (breakdown) and Anabolic (synthesis) pathways

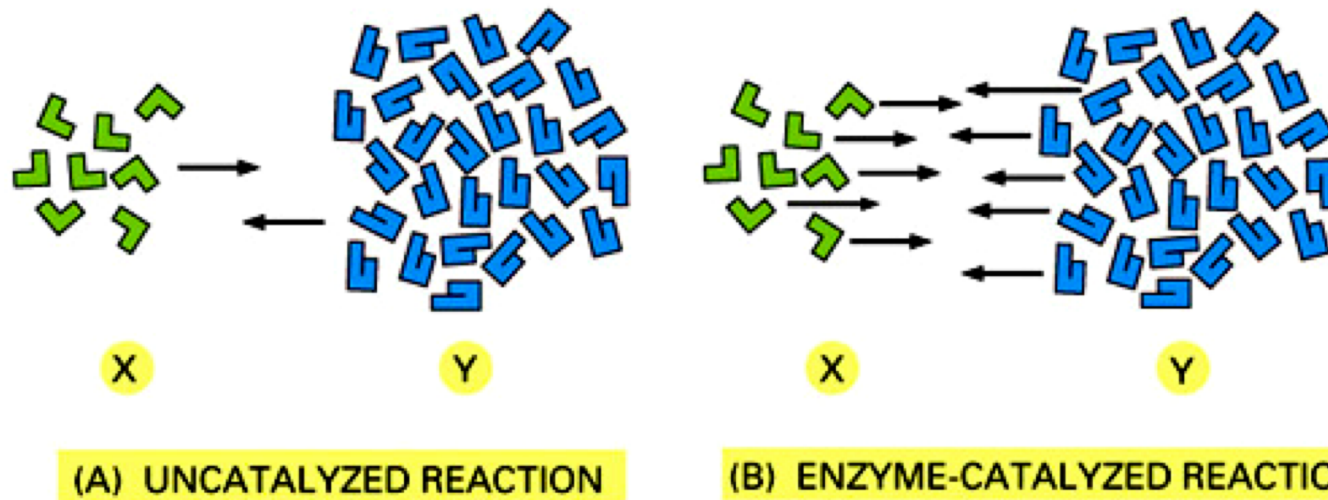


Enzymes catalyze metabolic reactions



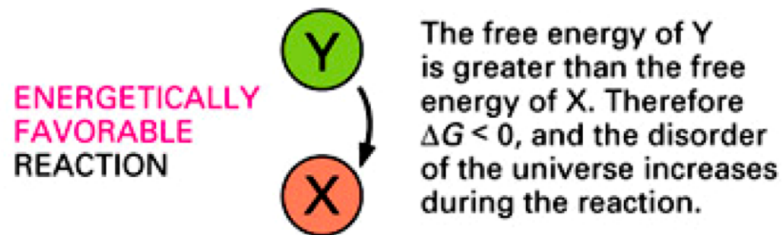
Enzymes accelerate reactions

-> *faster at equilibrium*

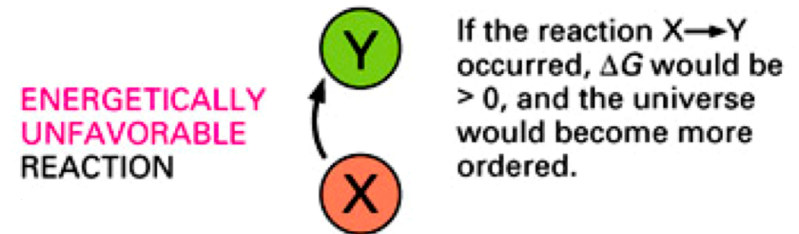


Enzymes can couple $\Delta G < 0$ reaction with $\Delta G > 0$ reaction

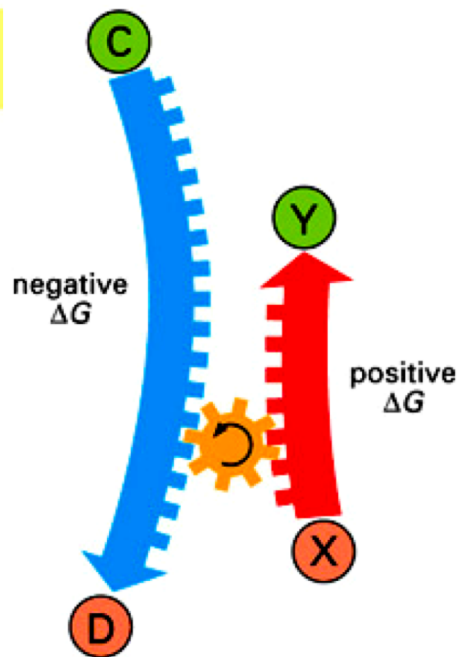
-> $\Delta G > 0$ becomes possible



this reaction can occur spontaneously

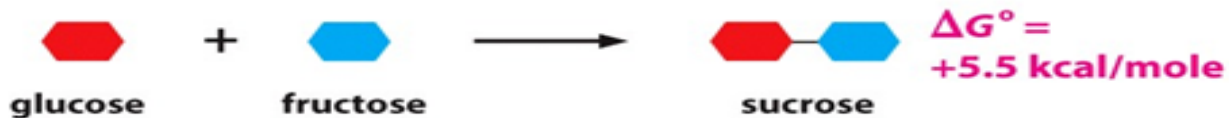


this reaction can occur only if it is coupled to a second, energetically favorable reaction



COUPLED REACTIONS

SINGLE REACTION

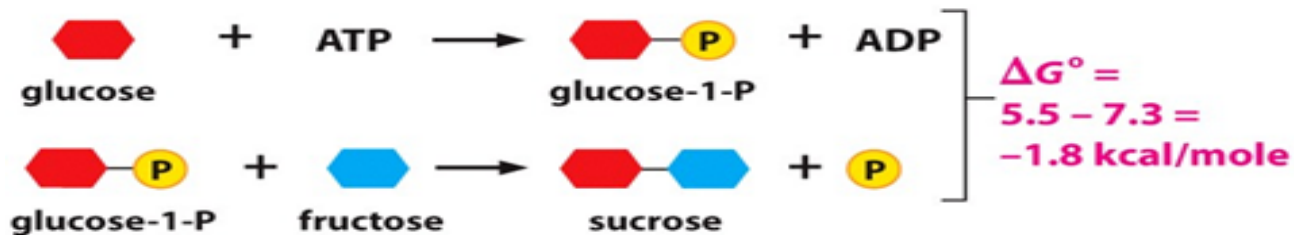


NET RESULT: reaction will not occur



NET RESULT: reaction is highly favorable

COUPLED REACTIONS



NET RESULT: sucrose is made in a reaction driven by the hydrolysis of ATP

Panel 3-1g Essential Cell Biology, 4th ed. (© Garland Science 2014)

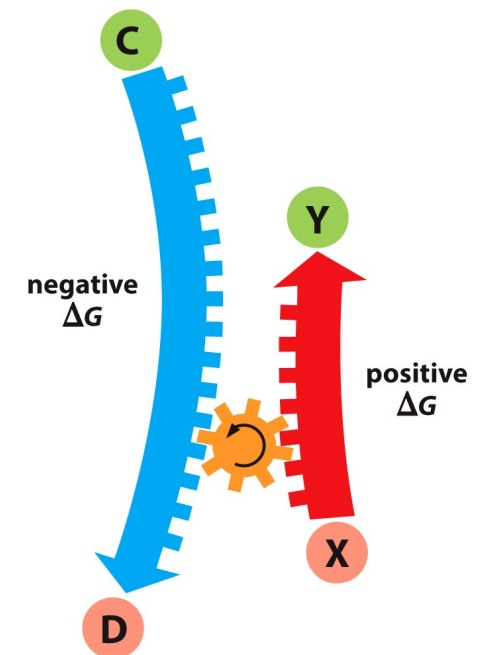
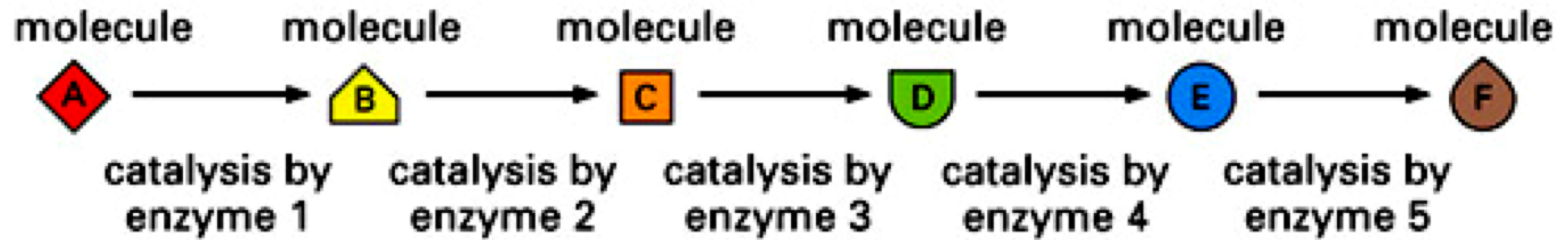


Figure 3-17 Essential Cell Biology, 4th ed. (© Garland Science 2014)

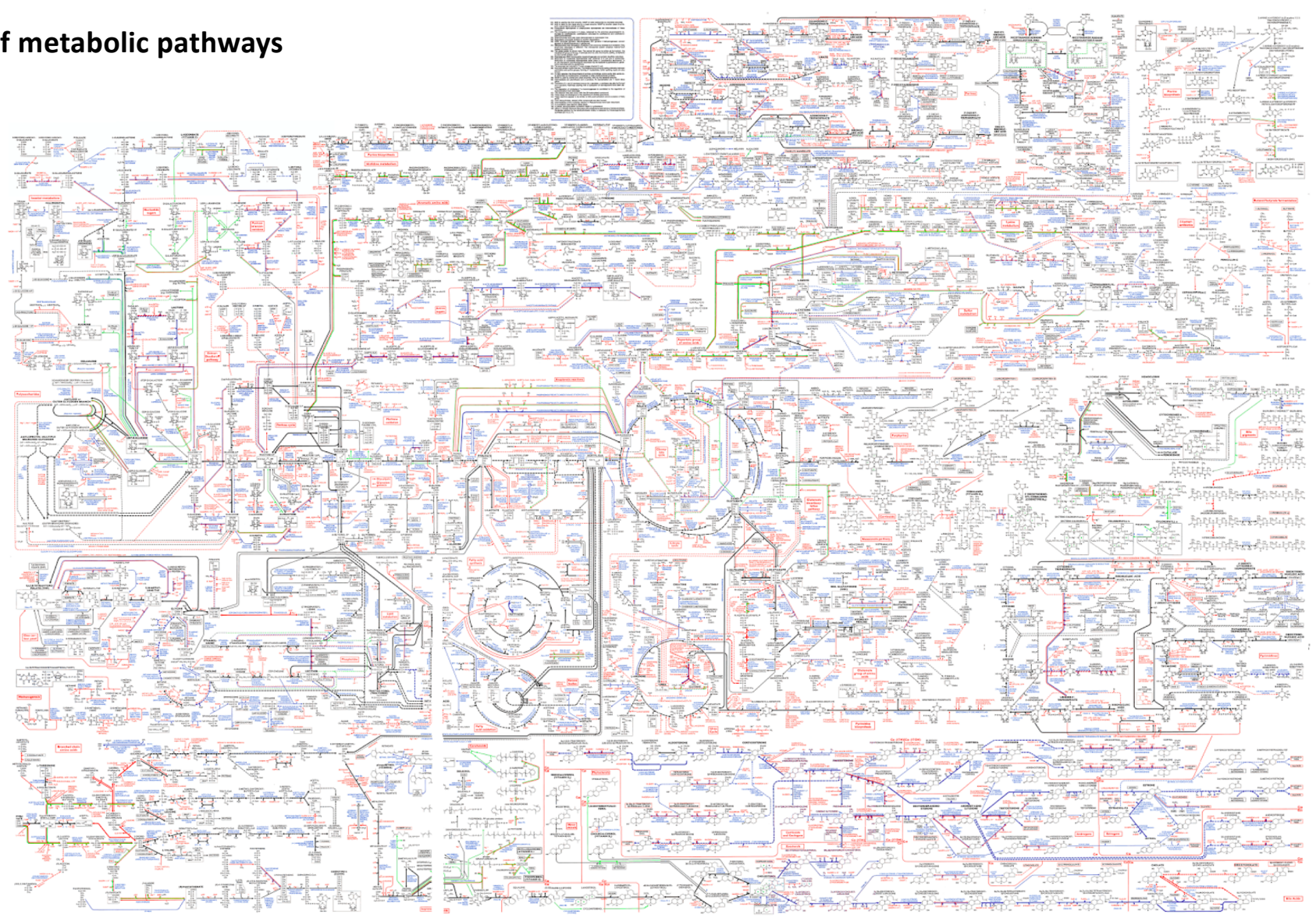
Metabolic pathways:



-> a chain of separate enzymatic reactions

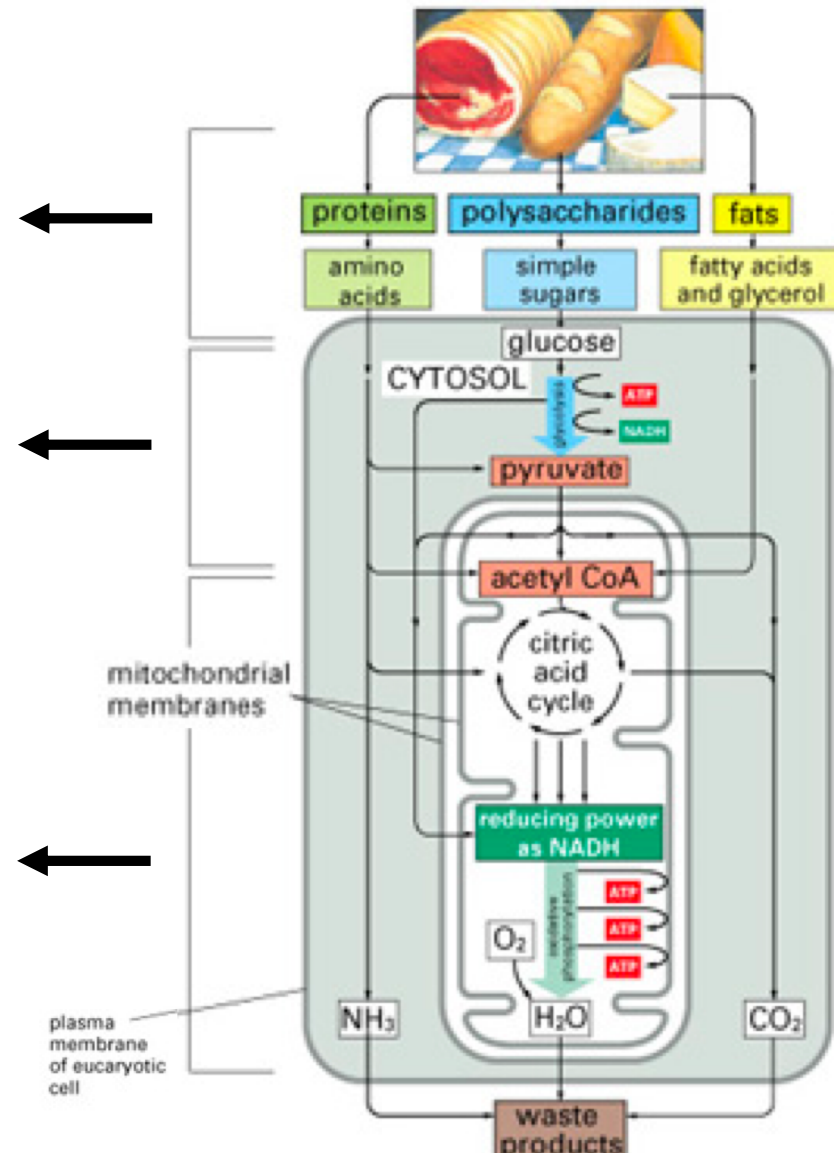
-> product of one reaction is substrate for next reaction

Network of metabolic pathways



How food ends up as H_2O , CO_2 and ATP

1. **breakdown of macromolecules** into their subunits (in stomach, intestines)
2. **breakdown of subunits** into intermediates Acetyl-Coenzyme A or pyruvate (in the **cytosol**)
3. **oxidation of Acetyl-Coenzyme A** into CO_2 and H_2O , with production of ATP (in **mitochondria**)



ATP is the universal energy carrier in the cell

-> metabolism of food must lead to ATP synthesis

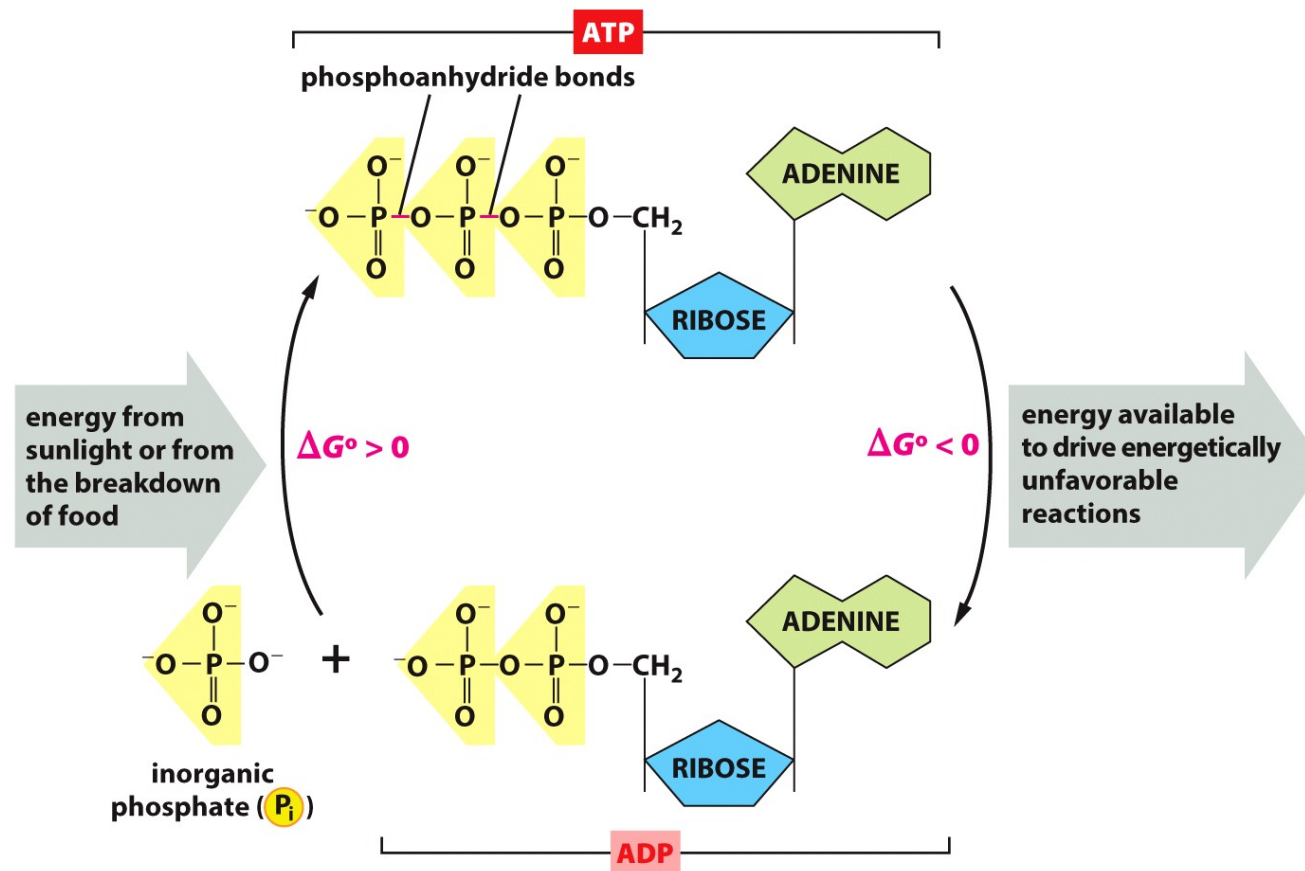
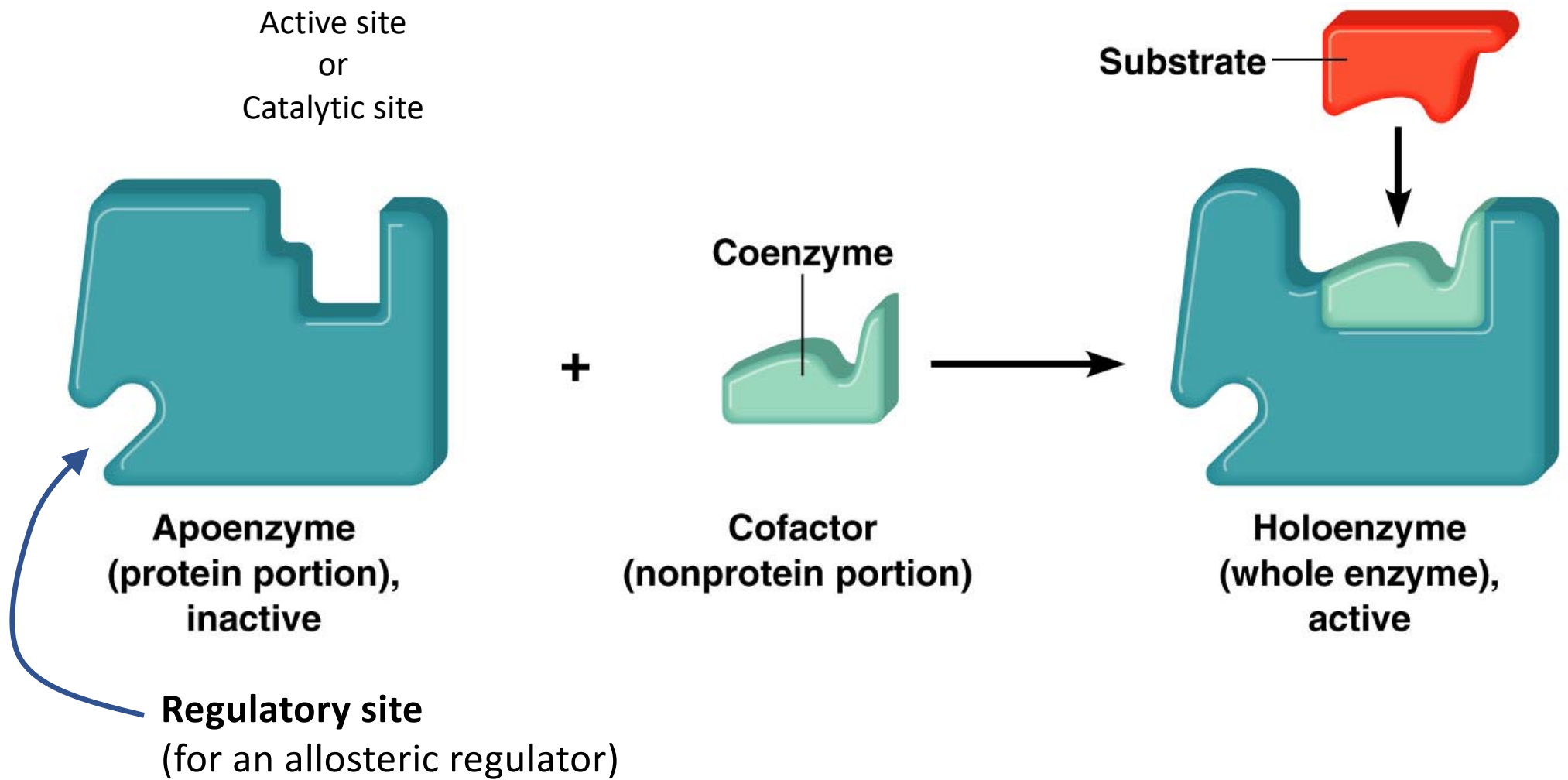


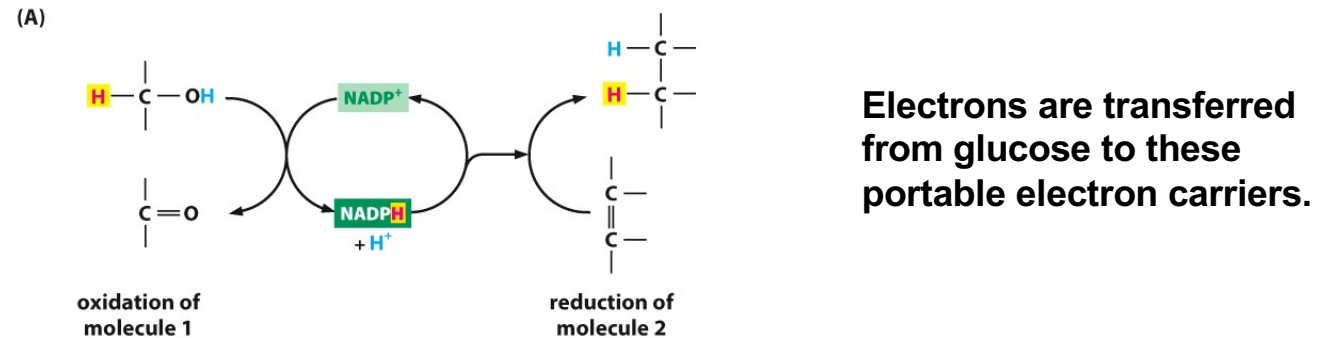
Figure 3-31 Essential Cell Biology, 4th ed. (© Garland Science 2014)

The concept of **coenzyme**



NADH and NADPH are Activated Carriers of Electrons

Some enzymes work with the help of a co-enzyme



co-enzymes :

$\text{NAD}^+ / \text{NADH} + \text{H}^+$

$\text{FAD}^+ / \text{FADH}_2$

CoA / Acetyl-CoA

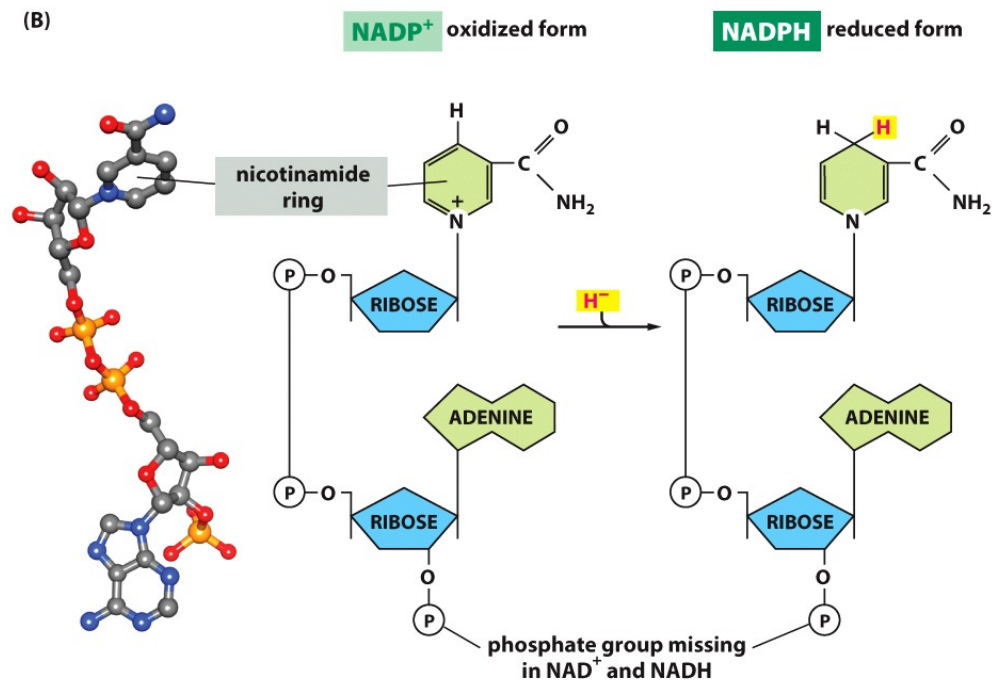
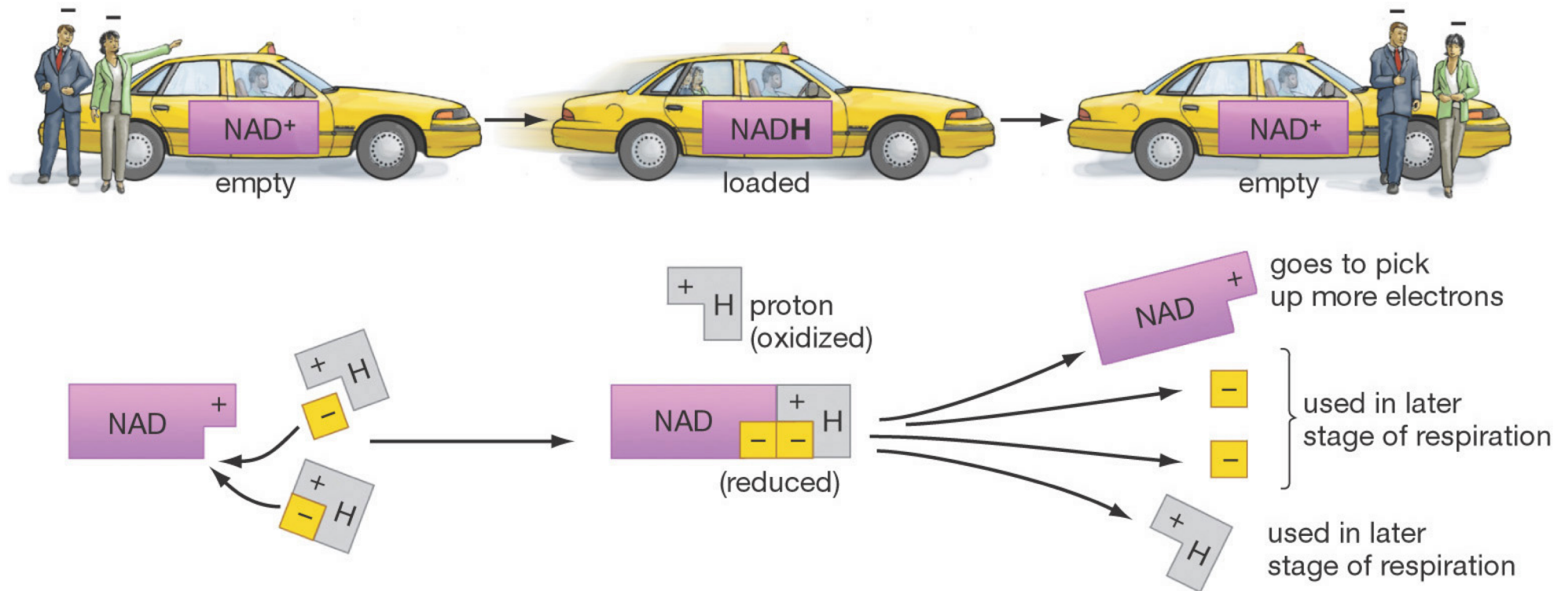


Figure 3-34 Essential Cell Biology, 4th ed. (© Garland Science 2014)

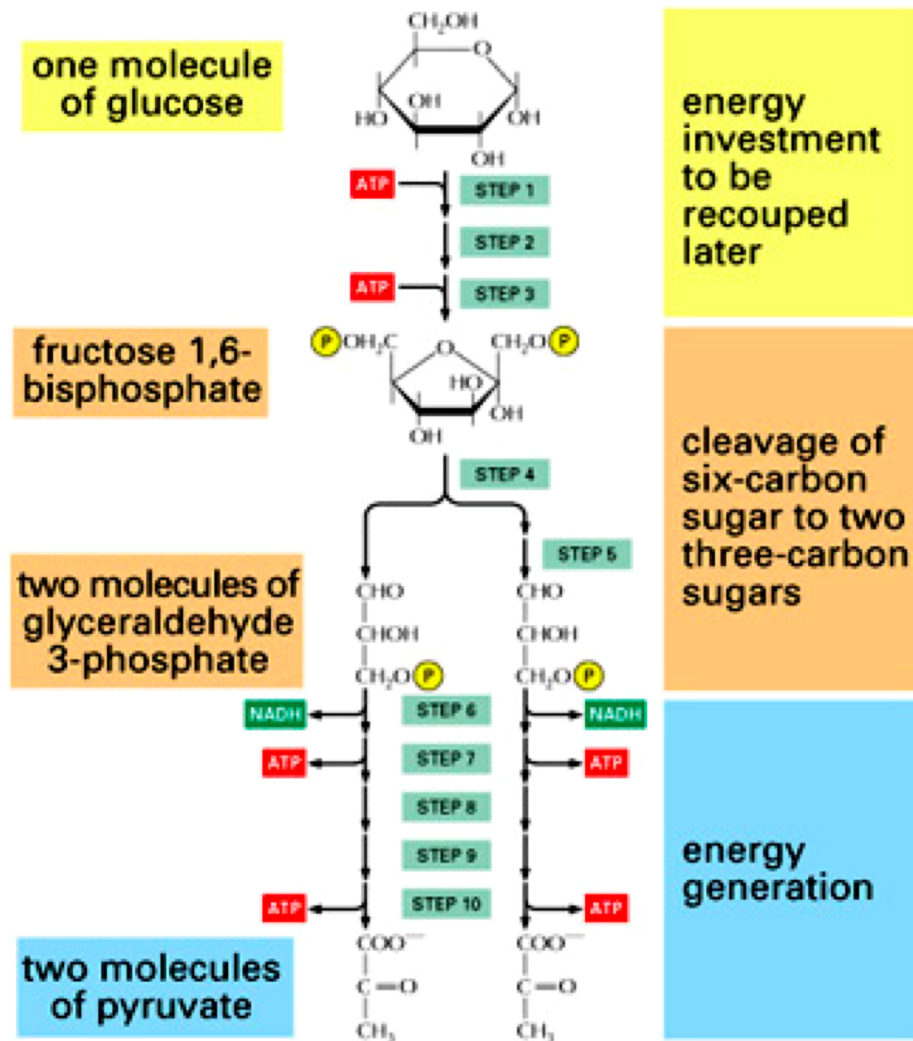
Coenzyme : an analogy



1. NAD^+ within a cell, along with two hydrogen atoms that are part of the food that is supplying energy for the body.

2. NAD^+ is reduced to NAD by accepting an electron from a hydrogen atom. It also picks up another hydrogen atom to become NADH .

3. NADH carries the electrons to a later stage of respiration then drops them off, becoming oxidized to its original form, NAD^+ .



Glycolysis

input: 1 glucose
2 ATP
2 NAD⁺

output: 2 pyruvate
4 ATP
2 NADH + H⁺